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$\frac{[CH_3COO^-][H^+]}{[CH_3COOH]}$

or with numbers $[H^+] =$

$\sqrt{(1.74 \times 10^{-5} \times$

$0.410)} = \sqrt{(7.13 \times 10^{-6})}$

$= 2.67 \times 10^{-3}$ -3 pH

$= 2.57$ can give three

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